# **Total, Asymmetric Synthesis of** ( + **)-Castanospermine,'**  ( + **)-6-Deoxycastanospermine, and** ( + **)-6-Deoxy-6-fluorocastanospermine**

Jean-Louis Reymond,<sup>t,2</sup> A. Alan Pinkerton,<sup>t</sup> and Pierre Vogel<sup>\*,t</sup>

*Section de chimie de l'Universit6 de Laurranne* CH-1005 *Lausanne, Switzerland, and Department of Chemistry, University of Toledo, Ohio* 

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Bromination of the dibenzyl acetal of  $(-)$ - $(1S,4S)$ -7-oxabicyclo $[2.2.1]$ hept-5-en-2-one  $((-)$ -5) led to  $(+)$ -**(1S,5S,6S,7S)-6-endo-(benzyloxy)-5-exo-bromo-7-oxabicyclo[** 2.2.lIheptan-2-one **(25).** Baeyer-Viier oxidation of 25 gave 2-O-benzyl-3-bromo-3,5-dideoxy- $\beta$ -L-arabino-hexofuranosidurono-6,1-lactone (26). Methanolysis of 26 afforded the corresponding methyl (methyl  $\alpha\beta$ -L-arabinofuranosid)uronates (27 + 28). The  $\alpha$  anomer 27 was reduced with DIBAH into methyl 2-O-benzyl-3-bromo-3,5-dideoxy- $\beta$ -L-arabino-hexofuranoside (29). Mesylatio reduced with DIBAH into methyl 2-*O*-benzyl-3-bromo-3,5-dideoxy-β-L-*arabino*-hexofuranoside (**29**). Mesylation<br>of the primary alcohol, followed by treatment with NH<sub>3</sub> gave methyl 2-*O*-benzyl-3,5,6-trideoxy-3,6-imino-βlyxo-hexofuranoside (32). Acetylation of the amine with CICH<sub>2</sub>COCl, acetolysis of the methyl furanoside followed by Arbuzov condensation with  $(EtO)<sub>3</sub>P$ , and then intramolecular Horner-Emmons reaction led to **(5S,6S,7S)-7-hydroxy-5-(benzyloxy)-l-azabicyclo[4.3.O]non-3-en-2-one (37).** Base-catalyzed hydrolysis of the corresponding epoxide 43 ((1S,6S,7S,8R,8aS)-8-(benzyloxy)-6,7-epoxy-1-hydroxyoctahydroindolizidin-5-one) followed by reduction of the lactam and deprotection of the alcoholic functions afforded (+)-castanospermine  $((+)$ -1). The conversion of **(-)-5** into **(+)-l** was highly stereoselective, requiring the isolation of 10 synthetic intermediates and with an overall yield of 15.2%. Reduction of 43 with BH<sub>3</sub>·Me<sub>2</sub>S or its treatment with HF-Et<sub>3</sub>N allowed one to prepare readily **(+)-6-deoxycastanospermine ((+)-2)** and **6-deoxy-6-fluorocastanospermine ((+)-3).** The crystal structure of **(+)-3** is also reported.

Castanospermine **((+)-l)** is a polyhydroxylated indolizidine isolated first in 1981 by Hohenschultz et al.<sup>3</sup> from seeds of the monotypic Australian rainforest and riverine tree *Castanospermum australe*. More recently,<sup>4</sup> it has been found in dried pod of *Alexa leiopetala* Sandwith and in seven other species of the same genus. The structure (relative configuration) of **(+)-l** was established by X-ray radiocrystallography first. $3$  and then (absolute configuration) through chemical correlation by Bernotas and Ganem.<sup>5</sup> Castanospermine has generated much interest because it is a potent inhibitor of various glucosidases including lysomal  $\alpha$ -glucosidase,<sup>6</sup>  $\alpha$ - and  $\beta$ -glucosidase in fibroplast extracts, $\frac{3}{5}$  the glycoprotein processing enzyme glucosidase **I7** as well as being a powerful inhibitor of **8**  xylosidase<sup>6</sup> and sucrase.<sup>8</sup> The ability of  $(+)$ -1 to disrupt glycoprotein processing has resulted in the use of this compound to modify biosynthesis; it might provide more insight into the role of oligosaccharides in glycoprotein function.<sup>9</sup> Very interesting is the fact that  $(+)$ -1 is able to inhibit experimental metastasis of some cancers.<sup>10</sup> Furthermore, it inhibits replication of human immunodeficiency virus (HIV) syncytium formation<sup>11</sup> and other virus replication.12



Ganem and Bernotas<sup>5</sup> have proposed a first synthesis of **(+)-l** which was inspired logically from the resemblance of **(+)-l** with D-glucopyranose. A short synthesis of 1 deoxynojirimycin (1,5-dideoxy-1,5-imino-p-glucitol) from

Dglucose was developed,13 and then **an** efficient two carbon chain elongation process (via the corresponding 6-carb-

(1) For a preliminary communication, see: Reymond, J.-L.; Vogel, P. *Tetrahedron Lett.* 1989, 30, 705. Enantiomerically pure 7-oxabicyclo- [2.2.l]hept-5-en-2-y1 derivatives ('naked sugars") **as** synthetic interme- diates, Part XIII, Part **XII,** see: Jeganathan, S.; Vogel, P. *J.* Org. *Chem.,*  in pre

(2) Present address: Research Institute of Scripps Clinic, 10666 North Torrey Pines Road, La Jolla, CA 92037.

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<sup>&#</sup>x27;Universit6 de Lausanne.

*<sup>t</sup>*University of Toledo.



aldehyde) was found to complete the synthesis of  $(+)$ -1.<sup>14</sup> The second approach proposed by Hashimoto and *co*workers<sup>15</sup> uses mannose as starting material. Fleet and co-workers<sup>16</sup> have obtained 6-epi- and 1,6-diepicastanospermine from L-gulonolactone, and Richardson and *co*workers derived 1-deoxycastanospermine from D-glucose<sup>17</sup> and **1,8-dideoxy-6-epicastanospermine** from a 2-azidoaltropyranoside derivative.<sup>18</sup> We report here a high stereoselective, total synthesis of **(+)-l** starting with (-)- **(1S,4S)-7-oxabicyclo[2.2.1]hept-5-en-2-one**  $(-)$ **-5), a "naked** sugar"<sup>19</sup>) obtained readily optically pure in two steps from furan and 1-cyanovinyl  $(1R')$ -camphanate  $(4).^{20,21}$  The first syntheses of **(+)-6-deoxycastanospermine ((+)-2)** and **(+)-6-deoxy-6-fluorocastanospermine ((+)-3)** are also reported. We also report the crystal structure of **(+)-3** by X-ray diffraction.

### **Retrosynthetic Plan**

We wished to develop a method that would allow us to prepare not only natural castanospermine **(+)-l,** but **also**  its enantiomer  $(-)$ -1 and derivatives in which some of the hydroxy groups could be exchanged by other substituents, with centers  $C(6)$ ,  $C(7)$ ,  $C(8)$ , and  $C(8a)$  conserving, or not, the gluco relative configuration. We have adopted a general strategy involving intermediate **9** with an unsaturated six-membered ring annulated to a pyrrolidine system (see Scheme II), the stereoselectivity of the substitution at  $C(6)$ and C(7) should be controlled by steric factors resulting from the geometry of the **l-azabicyclo[4.3.0]nonane** system or/and by the substituent at  $C(8)$ . In their synthesis of 8-episwainsonine, Austin and co-workers<sup>22</sup> showed that intramolecular Horner-Emmons reaction of the phos-





phonate **7** derived from pyrrolidine **6** by condensation with (MeO)<sub>2</sub>PO-CH<sub>2</sub>COOH gives a 47% yield of the corresponding lactam **8** after catalytic hydrogenation and acetylation (Scheme I).23

In analogy, the  $\alpha$ , $\beta$ -unsaturated lactam 9 should be derived from aldehyde **10** or an equivalent synthetic intermediate. The pyrrolidine ring in **10** would be generated through an intramolecular  $S_N2$  displacement of the 6**amino-5,6-dideoxy-arabino-hexose** derivative **11** in which the OH group at C(3) has been replaced by an adequate leaving group **X.** the furanose **12** form of **11** should be derived from the corresponding lactone  $13$  expected<sup>19,24</sup> to be formed with high regioselectivity in the Baeyer-Villiger oxidation of the 7-oxanorbornan-2-one **14** bearing at C(5) the leaving group  $X$  in the exo position and at  $C(6)$  a protected alcohol function in the *endo* position. This kind of compound can be obtained, in principle, in one step by

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**<sup>(16)</sup>** Fleet, G. W. J.; Ramsden, N. **G.;** Molyneux, R. J.; Jacob, **G.** S. *Tetrahedron Lett.* **1988,29,3603.** See also refs **13** and **14.** 

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electrophilic addition of **7-oxanorborn-5-en-2-one,** the electrophile attacking the exo face of the endocyclic double bond and the nucleophile being captured in the endo position of C(6) due to the electron-releasing effect of the homoconjugated carbonyl group.<sup>25,26</sup> Thus bromination of **(3-5** in an appropriate nucleophilic solvent ROH might lead to the desired derivative  $14$  in which  $X = Br$ .

#### **Results and Discussion**

Addition of  $\text{Br}_2$  to the racemic enone  $(\pm)$ -5<sup>24a</sup> in Ac<sub>2</sub>O + AcOH (-50 °C, 5 min) gave the unstable bromide  $(\pm)$ -17 in moderate yield. The latter arises from a Wagner-Meerwein (pinacolic) rearrangement of the bromonium ion intermediate  $(\pm)$ -15 into the oxycarbenium ion intermediate  $(\pm)$ -16,<sup>28</sup> which is then trapped with AcOH. Interestingly, and as in the case of the reaction of 5,6-exo-ep**oxy-7-oxabicyclo**[2.2.1]heptan-2-one in  $Ac_2O/HSO_3F/$ CH2Clz which gave mostly **5-oxo-2-oxabicyclo[2.2.l]hep**tane-2,7-diyl diacetates, the 1,2-shift of the acyl group in  $(\pm)$ -15 appears to be favored over that of the alkyl group.<sup>29</sup>



Addition of  $Br_2$  to  $(\pm)$ -5 in MeOH  $(-50 °C)$  to 20 °C, 15 h) gave bromohydrin  $(\pm)$ -18 (43%, isolated) together with products by decomposition. **This** result *can* be interpreted in **terms** of the mechanism shown in Scheme I11 involving the formation of a hemiacetal, due to MeOH addition to the ketone moiety of  $(\pm)$ -5, and bromination of the endocyclic double bond, leading to intermediate **(\*)-19.** The latter ion undergoes then 1,3-migration of the endohydroxy group via **(\*)-20** to generate onium ion intermediate ( $\pm$ )-21. Acid-catalyzed rearrangements of 3,8-dioxatricyclo [3.2.1.0<sup>2,4</sup>] octan-6-one<sup>29</sup> and 3-aza-8-oxatricyclo-[3.2.1.02\*4]octan-6-one acetals30 **(22)** given the corresponding **5-exo,6-endo-disubstituted** 7-oxanorbornan-2-ones **23** in high yield following similar mechanisms **as** that shown in Scheme III. Thus, the observation of  $(\pm)$ -5 + Br<sub>2</sub> + 2



**22 Z=O NCOR'; R=Me; CH2Ph=Bn 23** 

 $MeOH \rightarrow (\pm)$ -18 + HBr suggested that bromination of an acetal of  $(\pm)$ -5 might undergo a similar rearrangement and generate the desired protected bromohydrin in a good yield. This was indeed the case with the dibenzyl acetal **(-)-24** (obtained by treatment of optically pure enone  $(-)$ -5<sup>19</sup> with BnOSiMe<sub>3</sub> and CF<sub>3</sub>SO<sub>3</sub>SiMe<sub>3</sub> in CH<sub>2</sub>Cl<sub>2</sub>)<sup>29</sup> which reacted with  $Br_2$  in  $CH_2Cl_2$  at  $-90$  °C to give  $\overline{(+)}$ -25 isolated in 98% yield. The reaction with  $\text{Br}_2$ , and the

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quenching with aqueous NaHCO<sub>3</sub> had to be carried out at -90 "C to avoid concurrent decomposition and formation of benzyl bromide.



Baeyer-Villiger oxidation of **(+)-25** with mCPBA (metachloroperbenzoic acid) in  $CH_2Cl_2$  containing NaHCO<sub>3</sub> (20 "C) gave lactone **(+)-26 (95%).** Treatment of **(+)-26**  with MeOH and SOCl<sub>2</sub> led to a 4:1 mixture of the methyl furanosides **(-)-27** + **(+)-28** (100%) from which **(+)-28**  could be separated by crystallization and reequilibrated with  $(-)$ -27 by treatment with MeOH/SOCl<sub>2</sub>. Reduction of uronate **(-)-27** with diisobutylaluminium hydride **(DI-**BAL) in THF/toluene gave alcohol **(-)-29** quantitatively, the bromide being not reduced under these conditions. The corresponding methanesulfonate **(-)-30** (obtained **by**  treatment of  $(-)$ -29 with CH<sub>3</sub>SO<sub>2</sub>Cl/pyridine/CH<sub>2</sub>Cl<sub>2</sub>) with 24% NH<sub>3</sub> in EtOH/H<sub>2</sub>O 1:1 (45  $\rm{^6C}$ , 1 day) afforded the primary amine **31** which generated the pyrrolidine **(-)-32**  under the conditions of its formation (99% yield, 95% pure by 'H NMR). The methanesulfonate derived from the minor furanoside **(+)-28** was decomposed under these conditions!

Reaction of  $(-)$ -32 with ClCH<sub>2</sub>COCl in pyridine/CH<sub>2</sub>Cl<sub>2</sub> furnished the chloroacetamide **(+)-33** (79% based on **(-)-27).** The methyl furanoside moiety in **(+)-33** was transformed into the corresponding acetyl furanoside **34**  (88%) by treatment with  $\rm{Ac_2O/H_2SO_4}$  at 0 °C (2 h).<sup>23g</sup> Arbuzov reaction on  $34$   $\rm{(P(OEt)_3, 130\text{ }^oC)}$  gave the corresponding phsphonoacetamide **35,** which was not isolated and treated immediately with  $K_2CO_3$  in EtOH (20 °C, 3) days) giving product **37** which arose from the intramolecular Horner-Emmons condensation of the intermediate aldehyde **36.** Acetylation (Ac20/DMAP) of **37** afforded **(+)-38** (49% based on **(-)-27).** 



Addition of  $Br_2(AcOH, 20 °C)$  to  $(\pm)$ -38 gave dibromide **(44-39 as** major product of reaction. The same reaction in the presence of AgOAc (AcOH/Ac<sub>2</sub>O,  $10 °C$ )<sup>31</sup> afforded a 1.5:l mixture of monobromides **40** and **(+)-41.** As expected for bicyclo[4.3.0]nonene derivatives, the endo face

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**<sup>(31)</sup> Tung, C. C.; Speziale, A. J.; Frazier, H. W.** *J. Org. Chem.* **1968, 28, 1514.** 

of the C(6)-C(7) double bond in **(+)-38** is more sterically



hindered than its exo face, thus favoring the electrophilic attack that gives the bromonium ion intermediate **42.** It is possible also<sup>32</sup> that the allylic benzyloxy group at  $C(8)$ stabilizes the onium ion as shown with the hypothetical conformer 42'. In agreement with the Fürst-Plattner rule<sup>33</sup> which favors the formation of trans-diaxial products, nucleophilic quenching of **42** was expected to be favored at  $C(6)$  rather than at  $\tilde{C}(7)$ . This was indeed the case for Br, but the selectivity was not significant in the case of AcO-.



The mixture of protected bromohydrins  $40 + (+)$ -41 was methanolyzed (MeOH/SOCl<sub>2</sub>, 20<sup>°</sup>C) and then treated with a base  $(2-(tert-butylimino)-2-((diethylamino)imi$ **no)-1,3-dimethylperhydro-1,3,2-diazaphosphorine** on polystyrene (BEMP), CH3CN, 20°C) to give epoxide **43**  (75%), which could not be purified by column chromatography on silica gel. The latter was hydrolyzed in the presence of the same base  $(BEMP/CH_3CN/H_2O, 100 °C)$ and then acetylated (Ac20, DMAP) to yield lactan **(+)-44**  (42%, based on **(+)-38).** The nucleophile attack of C(6) was expected to be favored with respect to the C(7) attack on the basis of an electronic factor  $(\alpha$ -position to the amide moiety) and of steric hindrance due to the benzyloxy group at  $C(8)$  which retards the nucleophilic attack at  $C(7)$ . In this case, the Fürst-Plattner rule<sup>33</sup> would have predicted a preferred attack of center  $C(7)$  rather than  $C(6)$ .



Lactam 44 has the same configuration at  $C(1)$ ,  $C(6)$ , C(7), C(8), and C(8a) as in  $(+)$ -castanospermine  $((+)$ -1). This was established by its 360-MHz 'H NMR spectrum which showed typical vicinal coupling constants  ${}^{3}J(H-C(6),H-C(7)) = {}^{3}J(H-C(7),H-C(8)) = {}^{3}J(H-C(8),H-C(8a))$  $= 9.5$  Hz for axial protons of a chair cyclohexane conformation.<sup>34</sup> Reduction of the lactam with  $BH_{3}$ .Me<sub>2</sub>S (THF, 20 °C), followed by methanolysis (MeOH,  $\check{K_2}CO_3$ , 60 °C), afforded the partially protected castanoapermine derivative **45** (95%). Debenzylation  $(H_2/Pd-C)$  of **45** furnished  $(+)$ -1 in 97% yield. The physical and spectal data of **(+)-1** so



Figure **1.** Projections of crystal structures of **(+)-l** and **(+)-3.** 

obtained were identical with those of an authentic sample of natural (+)-castanospermine.

Treatment of epoxy lactam 43 with BH<sub>3</sub>·Me<sub>2</sub>S in anhydrous THF led to a mixture of compounds from which the partially protected 6-deoxycastanospermine **(+)-46**  could be isolated (25%). No other amine derivative was detected in the reaction mixture. Hydrogenolysis of the benzylic ether gave **(+)-2** (97%). Treatment of **43** with  $HF<sub>1</sub>$ Et<sub>3</sub>N led to a stereoselective ring opening of the epoxide moiety with attack of C(6) by the fluoride anion. After acetylation **(+)-47 (52%)** was isolated. Reduction of  $(+)$ -47 with  $BH_3$ ·Me<sub>2</sub>S in anhydrous THF, followed by hydrolysis of the acetates  $(HCl/MeOH/H<sub>2</sub>O, 70 °C)$  afforded the partially protected 6-deoxy-6-fluorocastanospermine **(+)-48** (83%). Hydrogenolysis of the benzylic ether gave **(+)-3** (93%).



The structures of  $(+)$ -2,  $(+)$ -3, and derivatives  $(+)$ -46 to **(+)-48** were secured by their spectral data and by their comparison with those reported for  $(+)$ -1.<sup>3,5,14,15</sup> For all these compounds, the 'H NMR spectra suggested conformations ( ${}^{N}C_{7}$  chair for the six-membered ring) similar to that of  $(+)$ -1.<sup>14,15</sup> The p $K_a$  values of the conjugate acids of **(+)-l, (+)-2,** and **(+)-3** have been determined (titrimetric method) to be  $6.01 \pm 0.01$ ,  $7.31 \pm 0.02$ , and  $5.09 \pm 0.01$ , respectively, at 25  $^{\circ}$ C (H<sub>2</sub>O).

Because the biological properties of **(+)-3** will have to be compared with those of **(+)-1,** we have determined the crystal structure of **(+)-3** by X-ray diffraction and compared it with that published for  $(+)$ -1.<sup>3</sup> As already suggested by the solution **'H** NMR spectra, the polyhydroxylated indolizidines (+)- **1** and **(+)-3** have similar topologies in the crystalline state **as** illustrated by their projections (Figure 1) in a plane perpendicular to the pseudo- $C_2$  axis of their five-membered rings passing through the carbon atom C(2) and the middle of the C- (8a)-N(4) bond (see also the supplementary material).

#### **Conclusion**

(+)-Castanospermine has been derived from (-)-7-oxa**bicyclo[2.2.1]hept-5-en-2-one ((4-5** in 15.2% overall yield). The method is highly stereoselective and requires the isolation of 10 intermediate products (see the experimental Section). The  $\alpha$ , $\beta$ -unsaturated lactam 37, as well as the corresponding epoxide **43,** are potential intermediates for the preparation of analogues of castanospermine **as** illustrated with the syntheses of 6-deoxy **((+)-2)** and 6 deoxy-6-fluor0 derivatives **((+)-3),** Since (+)-7-oxabicy-

**<sup>(32)</sup> Chamberlin, A. R.; Mulholland, R. L.; Kahn, S. D.; Hehre, W. J.** 

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 $\text{clo}[2.2.1]$ hept-5-en-2-one  $((+)$ -5) is as readily available as **(-)-5,lg** (-)-castanospermine and its derivatives **can** be prepared with the same ease.

### **Experimental Section**

General remarks, see ref 21.

(&)-( 1 RS,3SR ,4SR **,7SR)-7-Bromo-5-oxo-2-oxabicyclo-**   $[2.2.1]$ hept-3-yl Acetate  $((\pm)$ -17). Bromine  $(0.18 \text{ mL}, 3.45 \text{ mmol})$ was added dropwise in 5 min to a stirred solution of  $(±)$ -7-oxa $bicyclo[2.2.1]hept-5-en-2-one ((\pm)-5,0.38 g, 3.45 mmol)<sup>24a</sup> in Ac<sub>2</sub>O$ (4 mL) cooled to -60 °C. After the mixture was stirred at  $-60$ <sup>o</sup>C for 5 more min, NaHCO<sub>3</sub> (0.5 g) was added, and the mixture allowed to reach 0 °C. It was then poured into  $H_2O$  (100 mL) and extracted with  $\rm CH_2Cl_2$  (40 mL, three times). The solvent was evaporated and the Ac<sub>2</sub>O was eliminated by azeotropic distillation with toluene, in vacuo. The residue was recyrstallized from Et<sub>2</sub>O: 0.2 g (34%) of colorless crystals; mp 86-90  $^{\circ}$ C dec; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  6.64 (dd, J = 3.5, 1.0, HC(2-exo)), 4.80 (m,  ${}^3J$  = 2.5, 1.0, 0.5,  $\mathcal{U}(HC(2), HC(4)) = \mathcal{U}(HC(1), HC(4)) = 1.0, HC(4)),$ 4.34 (dd,  $J = 1.5, 0.5, \text{HC}(7)$ ), 3.20 (m,  ${}^3J = 3.5, 1.5, \frac{4J(\text{HC}(1))}{4J(\text{HC}(1))}$  $HC(5) = {}^{4}J(HC(1),HC(4)) = 1.0, HC(1)), 2.76$  (d,  ${}^{2}J = 18.5, {}^{3}J = 1.0, HC(5-endo)$ ), 2.48 (ddd,  ${}^{2}J = 18.5, {}^{3}J = 2.5, {}^{4}J = 1.0$ . HC(5-exo)), 2.09 (s, CH<sub>3</sub>CO)

**(f)-( 1 RS,2RS,3SR,4RS)-3-exo-Bromo-6,6-dimethoxy-7 oxabicyclo[2.2.l]heptan-2-endo-ol** ((\*)-18). A 1 M solution of Br<sub>2</sub> in anhydrous MeOH (1.4 mL) was added dropwise to a vigourously stirred solution of  $(\pm)$ -7-oxabicyclo[2.2.1] hept-5-en-2-one ((\*I-5, 150 mg, 1.4 mmol) in MeOH **(5** mL) cooled to -70 OC. The temperature was allowed **to** reach 20 "C overnight. The red solution was poured into a saturated aqueous solution of NaHCO<sub>3</sub>, and the mixture was extracted with  $CH_2Cl_2$  (40 mL, five times). After filtration on cotton and solvent evaporation, the residue was recrystallized twice from CH<sub>2</sub>Cl<sub>2</sub>/petroleum ether, yielding 150 mg (43%) of colorless crystals: mp 129-131 °C; IR (KBr)  $\nu$  3380, 2995, 2975, 2920, 1450, 1400; <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>)  $\delta$  4.61 (br d, J = 6.0, HC(4)), 4.57 (m, HC(2)), 4.41 (d, J  $= 10.5, OH$ , 4.36 (br d,  $J = 5.0, HC(1)$ ), 3.76 (d,  $J = 3.0, HC(3)$ ), 3.38 and 3.33 (2 s, 2 MeO), 2.15 (dd  $J = 13.0, 6.0, \text{HC}(5\text{-}exo)$ ), 1.97 (d,  $J = 13.0$ , HC(5-endo)); MS (CI, NH<sub>3</sub>) 223 (54), 222 (19), 221 (561, 220 (12), 173 (100).

 $(1S,4S,5S,6S)$ -6-endo-(Benzyloxy)-5-exo-bromo-7-oxa**bicyclo[2.2.1]heptan-2-one**  $((+)$ -25). A solution of  $Br_2$  (0.8 mL, 15.6 mmol) in anhydrous  $\text{CH}_2\text{Cl}_2$  (25 mL) was added dropwise (in 60 min) to a vigourously stirred solution of dibenzyl acetal of (-)-(1S,4S)-7-oxabicyclo[2.2.1]hept-5-en-2-one ((-)-24,<sup>19,29</sup> 4.5 g, 14.6 mmol) in anhydrous  $CH_2Cl_2$  (50 mL) cooled to -90 °C, under Ar atmosphere. A saturated aqueous solution of NaHCO<sub>3</sub> (20 mL) was added dropwise with stirring at -90  $^{\circ}$ C, and the mixture was allowed to warm up to  $0^{\circ}$ C. The mixture was then poured into saturated aqueous solution of  $\text{NaHCO}_3$  (150 mL) and extracted with  $CH_2Cl_2$  (30 mL, five times). After drying  $(MgSO_4)$ and solvent evaporation, the residue was crystallized from  $Et<sub>2</sub>O/petroleum ether, yielding 4.25 g (98%) of colorless crystals:$ mp 92-92.5 °C; <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>) δ 7.43-7.33 (m, C<sub>6</sub>H<sub>5</sub>), 4.89 (d,  $J = 6.5$ , HC(4)), 4.62 & 4.60 (2 d,  $J = 11.5$ , Bn), 4.48 (s, exo)), 2.29 (d,  $J = 18.0$ , HC(3-endo));  $[\alpha]^{20}$ <sub>589</sub> = +69°,  $[\alpha]^{20}$ <sub>578</sub> =  $g/dm^3$ ,  $CH_2Cl_2$ ). HC(6) + HC(l)), 4.01 *(8,* HC(5)), 2.61 (dd, *J* = 18.0,6.5, HC(3-  $+72^\circ$ ,  $[\alpha]_{\infty}^{20}$ ,  $[\alpha]_{\infty}^{8}$  = +81°,  $[\alpha]_{\infty}^{20}$ <sub>436</sub> = +129°,  $[\alpha]_{\infty}^{20}$ <sub>365</sub> = +160°  $(c = 10)$ 

*(&)-(-endo* -( Benzyloxy)-5-exo -bromo-7-oxabicyclo[ 2.2.11 heptan-2-one  $((\pm)$ -25). The same procedure was followed as for (+)-25, starting with **(f)-5,5-bis(benzyloxy)-7-oxabicyclo[2.2.1]**  hept-2-ene  $((\pm)$ -24),<sup>29</sup> mp 74–74.5 °C.

2- *0* **-Benzyl-3-bromo-3,5-dideoxy-&~-arabino** -hexo**furanosidurono-6,1-lactone** ((+)-26, (15,55,65,75)-7 endo - (Benzyloxy)-6-exo -bromo-2,8-dioxabicyclo[3.2.1]octan-3-one). NaHCO<sub>3</sub> (5 g) and then m-chloroperbenzoic acid (9.0 g, **Aldrich** 80-90%) were added to a stirred solution of (+)-25 (11.7 g, 39.4 mmol) in  $CH_2Cl_2$  (150 mL) cooled to 5 °C. After the mixture was stirred at 25 °C for 15 h, KF (4.7 g) was added, and the resultant mixture was stirred at 25 "C for 3 h. The precipitate was filtered off on Celite, and the solvent was evaporated. The residue was recrystallized from Et<sub>2</sub>O (100 mL) and petroleum ether (150 **mL),** yielding 11.8 g **(96%)** of colorless crystals: mp 115-116 °C; <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>)  $\delta$  7.37 (br s, C<sub>6</sub>H<sub>5</sub>), 5.90 (d, J =

4.0, HC(1)), 4.74 (d,  $J = 7.0$ , HC(5)), 4.70 & 4.60 (2 d,  $J = 11.5$ ,  $\text{Bn}$ ), 4.56 (d, J = 4.0, HC(7)), 4.08 **(s, HC(6))**, 3.13 **(dd, J** = 18.5,  $7.0, \text{HC}(4\text{-exo})$ , 2.69 (d,  $J = 18.5, \text{HC}(4\text{-endo})$ );  $\lbrack \alpha \rbrack^{20}$ <sub>580</sub> = +135<sup>o</sup>,  $[\alpha]^{20}$ <sub>578</sub> = +140°,  $[\alpha]^{20}$ <sub>548</sub> = +160°,  $[\alpha]^{20}$ <sub>436</sub> = +274°,  $[\alpha]^{20}$ <sub>365</sub> = +439°  $(c = 10 \text{ g/dm}^3, \text{CH}_2\text{Cl}_2).$ 

(±)-2-O-Benzyl-3-bromo-3,5-dideoxy-β-D,L-arabino-hexofuranosidurono-6,1-lactone  $((\pm)$ -26). The same procedure was followed as for  $(+)$ -26, starting with  $(\pm)$ -25: mp 107.5-108 °C.

Methyl (Methyl 2-O-benzyl-3-bromo-3,5-dideoxy-a-L**arabinofuran0sid)uronate** ((-)-27) and Methyl (Methyl 2- *0* - ben z y l-3- bromo-3,5 - dideox y *-B-~-ara bin* **o** - hexofuranosid)uronate  $((+)$ -28).  $SOL<sub>2</sub>(2 mL)$  was added dropwise into a stirred suspension of (+)-26 (4.2 **g,** 13.4 mmol) in MeOH (40 mL) at 25 °C. After having stood at 25 °C for 4 days, the mixture was cooled to  $0 °C$  and NaHCO<sub>3</sub> (7 g) was added portionwise in 10 min under vigourous stirring. The solvent was evaporated, the residue was dissolved in a saturated aqueous solution of NaHCO<sub>3</sub>, and the mixture was extracted with  $\text{CH}_2\text{Cl}_2$ (40 mL, five times). The extracts were combined and dried (MgS04), and the solvent was evaporated. The residue was crystallized form  $Et<sub>2</sub>O$  (30 mL) and petroleum ether (25 mL) at  $-20$  °C (3 days), yielding 0.78 g (16.2%) of a 1:5 mixture of (-)-27 and (+)-28. The mother liquor was evaporated and dried under vacuum, yielding 4.0 g (82.1%) of pure  $(-)$ -27. The crystalline mixture of  $(-)$ -27 +  $(+)$ -28 was dissolved in MeOH (7 mL) and treated as above with SOCl<sub>2</sub>. Separation by column chromatography on silica gel (Lobar B, ether,  $R_A(-)$ -27) = 0.6,  $R_A(+)$ -28)  $= 0.4$ ) gave 640 mg of (-)-27 and 120 mg of (+)-28. Global yield of  $(-)$ -27:4.64 g  $(97.8\%)$ .

Characteristics of  $(-)$ -27: colorless crystals; mp 32-35 °C; <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>)  $\delta$  7.42–7.31 (m, C<sub>6</sub>H<sub>5</sub>), 4.90 (d, J = 1.5,  $HC(1)$ , 4.71 and 4.63 (2 d,  $J = 12.0$ , Bn), 4.53 (ddd,  $J = 9.0$ , 8.5, 4.0, HC(4)), 4.21 (dd,  $J = 5.0$ , 1.5, HC(2)), 3.82 (dd,  $J = 9.0, 5.0$ , HC(3)), 3.74 (s, COOCH,), 3.38 **(8,** MeO), 2.83 (dd, J <sup>=</sup>16.0,4.0), and 2.63 (dd,  $J = 16.0$ , 8.5, H<sub>2</sub>C(5)); [a]<sup>20</sup><sub>589</sub> = -45<sup>o</sup>, [a]<sup>20</sup><sub>578</sub> =  $g/dm^3$ ,  $CH_2Cl_2$ ).  $-46.5^\circ$ ,  $[\alpha]^{20}$ <sub>546</sub> = -52.5°,  $[\alpha]^{20}$ <sub>436</sub> = -82°,  $[\alpha]^{20}$ <sub>365</sub> = -114°  $(c = 10)$ 

Characteristics of  $(+)$ -28: colorless crystals; mp 70-70.5 °C; <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>)  $\delta$  7.43–7.29 (m, C<sub>6</sub>H<sub>5</sub>), 4.78 and 4.75  $(2 d, J = 12.0, Bn)$ , 4.61 (d,  $J = 4.0$ , HC(1)), 4.56 (ddd,  $J = 9.5$ , 7.5, 4.0, HC(4)), 4.14 (dd,  $J = 9.5$ , 7.5, HC(3)), 4.08 (dd,  $J = 9.5$ , 4.0, HC(2)), 3.74 **(s,** COOCH,), 3.38 **(s,** MeO), 2.81 (dd, J <sup>=</sup>15.5, 4.0,  $\text{HC}(2)$ ,  $3.14$  (s, COOCH<sub>3</sub>), 3.58 (s, MeO), 2.61 (dd, J<br>4.0), and 2.58 (dd, J = 15.5, 9.5, H<sub>2</sub>C(5));  $\left[\alpha\right]_{889}$  = +89<sup>o</sup>,  $= +93^{\circ}, [\alpha]^{20}$ <sub>546</sub> = +105°,  $[\alpha]^{20}$ <sub>436</sub> = +174°,  $[\alpha]^{20}$ <sub>365</sub> = +265° (*c*)  $= 10 \text{ g/dm}^3$ ,  $\text{CH}_2\text{Cl}_2$ ).

(&)-Methyl (Methyl 2-0 **-benzyl-3-bromo-3,5-dideoxy-a-**D,L-arabino-hexofuranosid)uronate  $((\pm)$ -27) and  $(\pm)$ -Methyl (Methyl 2-*O* · benzyl-3-bromo-3,5-dideoxy-β-D,L-arabino · hexofuranosid)uronate  $((\pm)$ -28). The same procedure was followed as for  $(-)$ -27 +  $(+)$ -28, starting with  $(\pm)$ -26. Characteristics of  $(\pm)$ -27: colorless oil. Characteristics of  $(\pm)$ -28: colorless crystals; mp  $64-65$  °C.

Methyl 2-O-Benzyl-3-bromo-3,5-dideoxy- $\beta$ -L-arabinohexofuranoside  $((-)-29)$ . A 1.2 M solution of diisobutylaluminium hydride in toluene (60 mL, 72 mmol) was added in 10 min to a stirred solution of (-)-27 (10.75 **g,** 29.9 mmol) in anhydrous THF (100 mL) at -50 °C under Ar atmosphere. The temperature was allowed to reach -20 **"C** in 30 min, and the mixture was stirred for 5 h. Aqueous HCl (3 N, 100 mL) **was**  added under vigourous stirring, and after 5 min at 25 °C, a saturated aqueous solution of NH<sub>4</sub>Cl (100 mL) was added, and the mixture was extracted with AcOEt (100 mL, twice). The extracts were combined and dried  $(MgSO<sub>4</sub>)$ , and the solvent was evaporated, yielding 9.93 g (100%) of a colorless oil. **An** analytical sample of (-)-29 was, obtained by purification of 260 mg of this oil by column chromatography on silica gel (Lobar B, ether), yielding 240 mg (96%) of a colorless oil: 'H NMR (360 MHz, CDCl<sub>3</sub>)  $\delta$  7.44-7.31 (m, C<sub>6</sub>H<sub>5</sub>), 4.92 (d, J = 1.5, HC(1)), 4.72 and 4.63 (2 d,  $J = 12.0$ , Bn), 4.31 (ddd,  $J = 9.5, 9.0, 3.5, \text{HC}(4)$ ), 4.21 (dd,  $J = 5.0$ , 1.5 HC(2)), 3.85 (br t,  $J = 6.0$ , H<sub>2</sub>C(6)), 3.78 (dd, J <sup>=</sup>9.5.5.0. HC(3)). 3.38 **(8.** MeO). 2.21-2.06 and 1.93-1.83 (2 m.  $H_2C(5)$ ;  $[\alpha]_{889}^{\infty} = -47.5^{\circ}$ ,  $[\alpha]_{878}^{\infty}$ ,  $[\alpha]_{848}^{\infty} = -55^{\circ}$ ,  $[\alpha]_{488}^{\infty} = -87^{\circ}$ ,<br> $[\alpha]_{888}^{\infty} = -121.5^{\circ}$  (c = 10 g/dm<sup>3</sup>, CH<sub>2</sub>Cl<sub>2</sub>).

Methyl 2-O-Benzyl-3-bromo-3,5-dideoxy-6-O-(methyl**sulfonyl)-β-L-arabino-hexofuranoside** ((-)-30). CH<sub>3</sub>SO<sub>2</sub>Cl (2.8 mL, 36 mmol) was added dropwise to a stirred solution of crude

 $(-)$ -29 (9.93 g, 30 mmol) in anhydrous Et<sub>3</sub>N (5.0 mL, 36 mmol) and anhydrous  $CH_2Cl_2$  (60 mL) cooled to 0 °C. After stirring at 0 "C for 2.5 h, the mixture was poured into a mixture of ice (50 g) and a saturated aqueous solution of  $NAHCO<sub>3</sub>$  (100 mL). The mixture was extracted with  $CH_2Cl_2$  (50 mL, four times). The extracts were combined and dried  $(MgSO<sub>4</sub>)$ , and the solvent was evaporated to dryness in vacuo, yielding 12.63 g (100%) of a yellowish oil. An analytical sample of  $(-)$ -30 was obtained by purification of 240 mg by column chromatography on silica gel **(Lobar** B, ether), yielding 230 *mg* (96%) of a colorless oil: 'H **NMFt**  (360 MHz, CDCl<sub>3</sub>)  $\delta$  7.42–7.31 (m, C<sub>6</sub>H<sub>5</sub>), 4.89 (d, J = 1.5, HC(1), 4.69 & 4.61 (2 d,  $J = 12.0$ , Bn), 4.40 (m, H<sub>2</sub>C(6)), 4.24 (td,  $J =$ 9.0, 3.5, HC(4)), 4.21 (dd,  $J = 5.0$ , 1.5, HC(2)), 3.71 (dd,  $J = 9.0$ , 5.0, HC(3)), 3.25 **(s,** MeO), 3.05 *(8,* CH3SO2), 2.38-2.28 and 2.04-1.93 (2 m, H<sub>2</sub>C(5));  $[\alpha]_{589}^{20} = -43^{\circ}, [\alpha]_{578}^{20} = -44.5^{\circ}, [\alpha]_{546}^{20}$  $= -50^{\circ}$ ,  $[\alpha]^{20}$ <sub>436</sub>  $= -79.5^{\circ}$ ,  $[\alpha]^{20}$ <sub>365</sub>  $= -112.5^{\circ}$  (c = 11 g/dm<sup>3</sup>, CH<sub>2</sub>Cl<sub>2</sub>).

**Methyl 2-O-Benzyl-3,5,6-trideoxy-3,6-imino-β-L-lyxo-** mi hexofuranoside  $(-)-32$ ). A mixture of crude  $(-)-30$  (5.71 g, 14.1) mmol), EtOH (120 mL), and 24% aqueous  $NH<sub>3</sub>$  (120 mL) was heated to 45 "C for 24 h. After concentration to ca. 100 mL in vacuo, KOH (3 g) and ice (100 g) were added. The mixture was extracted with  $CH_2Cl_2$  (40 mL, five times). The extracts were combined, and the solvent was evaporated to dryness, yielding 3.48 g (99.4%) of a colorless oil. An analytical sample of  $(-)$ -32 was obtained by purification of 100 *mg* by column chromatography on neutral alumina (Et<sub>2</sub>O/MeOH, 10:1), yielding 65 mg (65%) of a colorless oil: <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\bar{\delta}$  7.38-7.30 (m,  $(m, Bn)$ , 3.92 (dd,  $J = 6.5$ , 1.5, HC(2)), 3.86 (dd,  $J = 6.5, 5.0$ , HC(3)), 3.34 *(8,* MeO), 3.10-2.92 (m, HzC(6)), 2.12 (br **s,** NH), 1.99  $(\text{ddd}, J = 13.5, 6.0, 2.5, 1.5)$ , and  $1.84$  (dddd,  $J = 13.5, 10.0, 7.5$ , 5.0, HzC(5)); *[alasee* <sup>=</sup>**-ao,** *[alss,\** = *-86",* **[alaw** = -97",  $= -154^{\circ}$ ,  $[\alpha]^{20}$ <sub>365</sub> =  $-218^{\circ}$  *(c = 10 g/dm<sup>3</sup>, CH<sub>2</sub>Cl<sub>2</sub>).*  $C_6H_5$ , 4.92 (d,  $J = 1.5$ , HC(1)), 4.68 (td,  $J = 5.0$ , 1.5, HC(4)), 4.64

Methyl 2- **O-Benzyl-N-(chloroacetyl)-3,5,6-trideoxy-3,6 imino-8-L-lyxo-hexofuranoside** (( +)-33). Anhydrous pyridine  $(2.26 \text{ mL}, 28 \text{ mmol})$  and then CICH<sub>2</sub>COCl  $(2.22 \text{ mL}, 28 \text{ mmol})$ were added dropwise to a stirred solution of  $(-)$ -32 (3.48 g, 14 mmol) in anhydrous  $CH_2Cl_2$  (50 mL) cooled to -5 °C. The temperature was allowed to reach 8 "C in 2 h with stirring. The mixture was poured into a mixture of ice  $(50 g)$  and  $1 N$  HCl $(120 g)$ mL) and extracted with  $CH_2Cl_2$  (30 mL, 6 times). Each extract was washed with saturated aqueous solution of NaHCO<sub>3</sub> (50 mL). The extracts were combined and the solvent was evaporated. The residue was purified by column chromatography on silica gel (150 g, AcOEt), yielding 2.59 g (79%) of a colorless oil: <sup>1</sup>H NMR (250) MHz, CDCl<sub>3</sub>) (3:1 mixture of two rotamers)  $\delta$  7.42-7.24 (m, C<sub>6</sub>H<sub>5</sub>), 4.934.98 **(8,** HC(l)), 4.83:4.80 (dd, J <sup>=</sup>6.5, 6.0, HC(3)), 4.54 **(s,**  2 H): 4.78 and 4.48 (2 d,  $J = 12.5$ , Bn), 4.12:3.94 (d,  $J = 6.0$ ,  $(\text{ddd}, J = 10.0, 8.0, 2.0, \text{HC}(6)), 3.48 \text{ (td, } J = 10.0, 7.0, \text{HC}(6)),$  $3.30:3.35$  (MeO),  $2.20-1.86$  (m,  $H_2C(5)$ );  $[\alpha]_{589} = +67.5^{\circ}$ ,  $[\alpha]_{578}$  $= +70.5^{\circ}$ ,  $[\alpha]_{\alpha}^{20} = +81.5^{\circ}$ ,  $[\alpha]_{\alpha}^{20} = +151^{\circ}$ ,  $[\alpha]_{\alpha}^{20} = +265^{\circ}$  *(c)*  $= 10 \text{ g/dm}^3$ ,  $\widetilde{\text{CH}_2\text{Cl}_2}$ ). HC(2)), 4.01 and 3.90:3.79 and 3.71 (2 d,  $J = 12.5$ , ClCH<sub>2</sub>CO), 3.68

Acetyl 2-O-Benzyl-3,5,6-trideoxy-3,6-imino-α-L-lyxohexofuranoside  $(34\alpha)$  and Acetyl 2-O-Benzyl-3,5,6-trideoxy-3,6-imino- $\beta$ -L-lyxo-hexofuranoside (34 $\beta$ ). Concentrated  $H<sub>2</sub>SO<sub>4</sub>$  (1.5 g) was added dropwise to a stirred solution of (+)-33  $(2.94 \text{ g}, 9.02 \text{ mmol})$  in Ac<sub>2</sub>O (30 mL) cooled to 0 °C. After the mixture was stirred at 5 °C for 2 h,  $NaHCO<sub>3</sub>$  (3 g) was added portionwise. After 5 min at 0 °C, the mixture was poured into ice  $(50 \text{ g})$  and a saturated aqueous solution of  $\text{NaHCO}_3$   $(100 \text{ mL})$ and extracted with  $CH_2Cl_2$  (70 mL, four times). Each extract was washed with brine (50 mL). The extracts were combined, the solvent was evaporated, and the excess of  $Ac_2O$  was eliminated by azeotropic distillation with toluene (300 mL, then 100 **mL,**  twice, rotatory evaporator). The yellowish residue was purified by column chromatography on silica gel (AcOEt), yielding 3.59 g (87.5%) of a yellowish oil that can be used in the next synthetic step. Analytical samples of  $34\alpha$  and  $34\beta$  were obtained on separating 150 mg by column chromatography on silica gel (Lobar B, AcOEt), yielding 112 mg (74%) of  $34\alpha$  and 11 mg (7%) of  $34\beta$ . characteristics of 34a: yellowish oil; 'H NMR (250 **MHz,** CDC13) (major rotamer) 6 7.40-7.27 (m, CsH6), 6.23 *(8,* HC(l)), 4.90-4.80  $(m, HC(3) + HC(4))$ , 4.64 and 4.54 (2 d,  $J = 11.0$ , Bn), 4.22 (d,  $J = 6.0$ , HC(2)), 4.02 and 3.90 (2 d,  $J = 12.5$ , ClCH<sub>2</sub>CO), 3.68 (ddd,  $J = 10.5, 8.0, 2.5$  and 3.49 (ddd,  $J = 10.5, 10.4, 7.0, H<sub>2</sub>C(6)$ ),

2.25-2.00 (m, HzC(5)), 2.07 *(8,* Ac); (minor rotamer) 6 6.29 **(8,**  HC(1)), 4.04 (d, J = 6.0, HC(2), 3.67 (s, ClCH<sub>2</sub>CO), 2.12 (s, Ac);  $[\alpha]_{0.589}^{20} = +65^{\circ}, [\alpha]_{0.578} = +68^{\circ}, [\alpha]_{0.546}^{20} = +78^{\circ}, [\alpha]_{0.486}^{20} = +144^{\circ},$  $= +250^{\circ}$  ( $c = 10$  g/dm<sup>3</sup>, CH<sub>2</sub>Cl<sub>2</sub>). Characteristics of 34 $\beta$ : yellowish oil; 'H NMR (250 MHz, CDC13, 2 rotamers **1:l)** 6 7.40-7.20 (m,  $C_6H_5$ ), 6.32 and 6.22 (2 d,  $J = 3.5$ , HC(1)), 4.97-4.47  $(m, 4$  H, HC(3), HC(4), Bn), 4.36 and 3.98 (2 d,  $J = 12.5$ , CICH<sub>2</sub>CO), 4.14-3.35 (m, HC(2), H<sub>2</sub>C(6), CICH<sub>2</sub>CO), 2.22-1.88 (m, HzC(5)), 2.10 and 2.08 (2 **s,** Ac).

 $(5S, 6S, 7S)$ -7-Acetoxy-5- $(benzyloxy)$ -1-azabicyclo[4.3.0]non-3-en-2-one ((+)-38). A mixture of  $34\alpha + 34\beta$  (1.57 g, 4.44 mmol) and triethyl phosphite (8 mL) was heated to 130 $\degree$ C for 7 h. The excess of  $(EtO)<sub>3</sub>P$  was distilled off under vacuum, and the residue was dried in vacuum (1 Torr, 130 "C, 15 min). After addition of EtOH (50 mL) and  $K_2CO_3$  (1.3 g), the mixture was stirred at 20 °C for 3 days. Water (100 mL) was added, and the mixture was extracted with  $CH_2Cl_2$  (200 mL, and then 50 mL, six times). The extracts were combined and dried **(MgS04),** and the solvent was evaporated. Anhydrous pyridine (5 mL), Ac<sub>2</sub>O (5 **mL),** and **4-(dimethylamin0)pyridine** (10 *mg)* were added, and the mixture was allowed to stand at 20 "C for 2 days. The excess of reagents was distilled off by azeotropic distillation with toluene (100 mL, then 50 mL, twice, rotatory evaporator), and the residue was purified by column chromatography on silica gel (100 g, AcOEt). The main fraction  $(R_f = 0.3)$  yielded 960 mg (72%) of a colorless oil: <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>)  $\delta$  7.40-7.28 (m, C<sub>6</sub>H<sub>5</sub>), 6.62 (dd,  $J = 10.0, 1.5, HC(4)$ ), 5.92 (dd,  $J = 10.0, 2.0, HC(3)$ ), 5.44 (ddd,  $J = 3.6, 3.5, 3.4, \text{HC}(7)$ ), 4.68 and 4.50 (2 d,  $J = 12.0$ , Bn), 4.46 (ddd,  $J = 11.5, 2.0, 1.5, \text{HC}(5)$ ), 3.86 (dd,  $J = 11.5, 3.6$ , HC(6)), 3.74 (ddd,  $J = 11.0$ , 9.5, 1.5) and 3.48 (ddd,  $J = 11.0$ , 10.9, 7.5, H<sub>2</sub>C(9)), 2.20–1.96 (m, H<sub>2</sub>C(8)), 1.88 (s, Ac); UV (CH<sub>3</sub>CN)  $\lambda$ 7.5, H<sub>2</sub>C(9)), 2.20–1.96 (m, H<sub>2</sub>C(8)), 1.88 (s, Ac); UV (CH<sub>3</sub>CN)  $\lambda$ <br>210 ( $\epsilon$  16 000), 220 (7200), 230 (1660), 240 (1330), 250 (1700),  $\lambda_{\text{max}}$ <br>257 (1800), 270 (1300), 280 (840), 300 (180); [ $\alpha$ ]<sup>25</sup><sub>589</sub> = +161°,  $= +328^{\circ}$  $(c = 10 \text{ g}/\text{dm}^3, \text{CH}_2\text{Cl}_2).$  $= +166^{\circ}, [\alpha]_{546}^{\infty} = +187^{\circ}, [\alpha]_{436}^{\infty} = +281^{\circ},$ 

(5RS ,6RS,7RS)- **l-Acetoxy-5-(benzyloxy)-l-azabicyclo-**   $[4.3.0]$ non-3-en-2-one  $((\pm)$ -38). The same procedure was followed as for  $(+)$ -38, starting with the mixture of  $(\pm)$ -27 and  $(\pm)$ -28: colorless crystals; mp 91.5-92.5 "C.

(1 RS ,6RS,7SR ,8SR,8aSR)- l-Acetoxy-8-( benzyloxy)-6,7 dibromooctahydroindolizidin-5-one ((±)-39). A 1 M solution of Br<sub>2</sub> in AcOH (0.4 mL, 0.4 mmol) was added dropwise to a stirred solution of  $(\pm)$ -38 (30 mg, 0.1 mmol) in AcOH (0.3 mL). After 3 h at 20 °C,  $H<sub>2</sub>O$  (50 mL) was added and the mixture extracted with  $CH_2Cl_2$  (30 mL, three times). The extracts were combined, the solvent was evaporated, and the excess of AcOH was distilled off by azeotropic distillation with toluene. The residue was crystallized from  $Et_2O$ , yielding 28 mg (61%) of colorless crystals: mp 162-168 °C dec; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) δ 7.40-7.30 (m,  $C_6H_6$ , 5.40 (q,  $J = 2.5$ , HC(1)), 4.80 (m, HC(6), HC(7)), 4.68 and 4.16 (2 d,  $J = 12.5$ , Bn), 4.20 (dd,  $J = 9.0$ , 2.0, HC(8)), 3.94 (dd,  $J = 9.0, 2.5, \text{HC}(8a)$ , 3.80-3.50 (m, H<sub>2</sub>C(3)), 2.18-2.05 (m, H<sub>2</sub>C(2)), 1.86 **(s,** Ac).

**(1** S ,6R ,7R ,8R ,8aS)-8-( Benzyloxy)-7-bromo- 1,6-diacet**oxyoctahydroindolizidin-5-one** (40) and (1s ,6R ,7S,8R,8aS)-8-( **Benzyloxy)-6-bromo-l,7-diacetoxyoctahydroindolizidin-5-one** ((+)-41). A freshly prepared 1 M  $Br<sub>2</sub>$  solution in Ac<sub>2</sub>O (3 mL) was added dropwise in 10 min to a stirred suspension of (+)-38 (300 mg, 1 mmol) and AgOAc (600 mg, 3.6 mmol) in AcOH (3 mL) and Ac<sub>2</sub>O (1.5 mL) at 9  $^{\circ}$ C and under Ar atmosphere. After stirring at 9 °C for 15 min, the mixture was poured into  $H_2O$  (100 mL) and ice (50 g) and extracted with AcOEt (100 mL, three times). The extracts were combined, washed with  $H<sub>2</sub>O$  (80 mL, three times), and dried (MgSO<sub>4</sub>), the solvent was evaporated, and the excess of Ac<sub>2</sub>O was distilled off by azeotropic distillation with toluene (20 mL, then 4 mL, four times, rotatoray evaporator), yielding 700 mg of a yellowish oil, 1.5:1 mixture of  $40 + (+)$ -41 that can be used for the next step. Analytical samples of 40 and (+)-41 were obtained in the following way. A 200-mg portion of  $40 + (+) \cdot 41$  was purified by column chromatography on silica gel (20 g, AcOEt/petroleum ether, 1:1,  $R_f = 0.5$ ), yielding 140 mg (70%) of a colorless oil that crystallized on adding  $Et_2O$  (4 mL, 20 °C), giving 40 mg (20%) of (+)-41. Evaporation of the mother liquor yielded a 51 mixture of  $40/$ (+)-41. Characteristics of 40:  $^{1}$ H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  7.40–7.25 (m, C<sub>6</sub>H<sub>5</sub>), 5.72 (d,  $J = 2.8$ , HC(6)), 5.44 (ddd,  $J =$ 

3.0, 2.5, 2.4, HC(1)), 4.66 and 4.45 (2 d,  $J = 12.0$ , Bn), 4.49 (dd,  $J = 2.8, 2.5, \text{HC}(7)$ ), 3.94 (dd,  $J = 8.5, 3.0, \text{HC}(8a)$ ), 3.70 (dd,  $J = 8.5, 2.5, \text{HC}(8)$ ), 3.80-3.52 (m, H<sub>2</sub>C(3)), 2.18-2.10 (m, H<sub>2</sub>C(2)), 2.08 and 1.93 (2 s, 2 Ac). Characteristics of (+)-41: colorless crystals; mp 126-127 OC; 'H NMR (250 MHz, CDCls) **6** 7.40-7.25  $(m, C_6H_5)$ , 5.58 (dd,  $J = 6.5, 6.4, HC(7))$ , 5.38 (ddd,  $J = 3.5, 3.4$ , 1.5, HC(1)), 4.68 and 4.52 (2 d,  $J = 11.5$ , Bn), 4.35 (d,  $J = 6.4$ , HC(6)), 3.91 (dd,  $J = 9.0$ , 3.5, HC(8a)), 3.76 (dd,  $J = 9.0$ , 6.5 HC(8)), 3.70-3.55 (m,  $H_2C(3)$ ), 2.20-2.07 (m,  $H_2C(2)$ ), 2.12 and  $1.93 (2 \text{ s}, 2 \text{ Ac})$ ;  $[\alpha]^{26}$ <sub>599</sub> =  $+123^\circ$ ,  $[\alpha]^{26}$ <sub>578</sub> =  $+128^\circ$ ,  $[\alpha]^{26}$ <sub>548</sub> =  $+146^\circ$ ,  $= +252^{\circ}, [\alpha]^{25}_{365} = +401^{\circ}$  *(c = 10 g/dm<sup>3</sup>, CH<sub>2</sub>Cl<sub>2</sub>).* 

**(1 S** ,6S ,7S ,8R,8aS)-8- (Benzyloxy)-6,7-epoxy- 1-hydroxy**octahydroindolizidin-5-one (43).** SOCl<sub>2</sub> (0.5 mL) was added dropwise to a stirred solution of the crude 1.5/1 mixture of 40  $+ (+) - 41$  (derived from 300 mg of  $(+) - 38$  (1 mmol)) in MeOH (10) mL) under Ar. After stirring at 20 "C for 17 h, the solvent was evaporated and the residue was dissolved in  $CH<sub>3</sub>CN$  (10 mL). After addition of **2-(tert-butylimino)-2-(diethylamino)-l,3-dimethylperhydro-1,3,2-diazaphosphorine** on polystyrene (900 mg, 2 mmol), the suspension was stirred at 20 "C for 35 min. The polymer supported base was filtered off, and the solvent was evaporated, giving an oily residue. An analytical sample of **43**  was obtained by purifying 40 mg by column chromatography on silica gel (1 g,  $\text{AcOEt}, R_f = 0.3$ ), yielding, 20 mg (50%) of a colorless oil: <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) δ 7.45-7.36 (m, C<sub>6</sub>H<sub>5</sub>), 4.86 and 4.64 (2 d,  $J = 12.0$ , Bn), 4.35 (m, HC(1)), 3.88 (dd,  $J = 10.0, 0.7$ , HC(8)), 3.76 (dd,  $J = 10.0$ , 3.0, HC(8a)), 3.66 (ddd,  $J = 11.5$ , 8.5, 2.0, HC(3)), 3.56 (dd, J <sup>=</sup>3.5, 0.7, HC(7)), 3.43 *(J* = 11.5, 7.5, H'C(3)), 3.38 (d,  $J = 3.5$ , HC(6)), 2.04-1.88 (m, H<sub>2</sub>C(2)), 1.54 (br d,  $J = 4.0,$  HO.

**(1S,6R,7S,8R,8aS)-8-(Benzyloxy)-l,6,7-triacetoxyocta**hydroindolizidin-6-one (( **+)-44).** The crude epoxide **43** (before chromatography) resulting from the transformation of 300 mg of  $(+)$ -38  $(1 \text{ mmol})$  was dissolved in CH<sub>3</sub>CN  $(2 \text{ mL})$  and H<sub>2</sub>O  $(16 \text{ m})$ mL). After the addition of **2-(tert-butylimino)-2-(diethylamino)-l,3-dimethylperhydro-1,3,2-diazaphosphorine** on polystyrene **(450** mg), the suspension was stirred at 100 "C for 4.5 h. The polymer was filtered off, and the solvent was evaporated. The residue was dissolved in anhydrous pyridine  $(4 \text{ mL})$  and  $Ac_2O$ (4 mL), and **4-(dimethylamino)pyridme** (10 *mg)* was added. After 2 days at 20 "C, the solvent was evaporated, and the excess of pyridine and Ac<sub>2</sub>O was distilled off by azeotropic distillation with toluene (70 mL, then 15 mL, twice, rotatory evaporator). The residue was taken with **0.5** N HCl (20 mL) and extracted with  $CH<sub>2</sub>Cl<sub>2</sub>$  (10 mL, four times). The extracts were combined, and the solvent was evaporated. The residue was purified by column chromatography on silica gel  $(3 g, AcOEt)$ , yielding 250 mg  $(R_f)$  $= 0.5$ ) of a colorless oil which crystallizes from Et<sub>2</sub>O, giving 174 mg (42%, based on **(+)-38)** of colorless crystals: mp 97-98 "C; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  7.40–7.20 (m, C<sub>6</sub>H<sub>5</sub>), 5.62 (t, J = 9.5, HC(7)), 5.38 (m, HC(1)), 5.10 (d,  $J = 9.5$ , HC(6)), 4.66 and 4.54 (2 d,  $J = 11.0$ , Bn), 3.88 (t,  $J = 9.5$ , HC(8)), 3.74 (dd,  $J =$ 9.5, 3.0, HC(8a)), 3.60 (m,  $H<sub>2</sub>C(3)$ ), 2.22-2.00 (m,  $H<sub>2</sub>C(2)$ ), 2.12, 2.06, and 1.98 (3 s, 3 Ac);  $[\alpha]^{\mathfrak{D}}_{\mathfrak{g}_{\mathfrak{B}}} = +136^{\circ}$ ,  $[\alpha]^{\mathfrak{D}}_{\mathfrak{g}_{\mathfrak{R}}} = +142^{\circ}$ ,  $[\alpha]^{\mathfrak{D}}_{\mathfrak{g}_{\mathfrak{K}}} = +152^{\circ}$ ,  $[\alpha]^{\mathfrak{D}}_{\mathfrak{g}_{\mathfrak{K}}} = +280^{\circ}$ ,  $[\alpha]^{\mathfrak{D}}_{\mathfrak{M}} = +451^{\circ}$  ( $c = 10 \text{ g$ 

**(1RS,6SR,7RS,8SR,8aRS)-8-(Benzyloxy)-1,6,7-triacetoxyoctahydroindolizidin.5-one** ((±)-44). The same procedure was followed as for  $(+)$ -44, starting with  $(+)$ -38: colorless crystals; mp 198-199 "C.

**(lS,6S,7R,8R,8aR)-8-(Benzyloxy)-l,6,7-trihydroxyocta**hydroindolizidine (45). A 10 M solution of BH<sub>3</sub>·Me<sub>2</sub>S in THF (0.6 mL, Aldrich) was added dropwise to a stirred suspension of (+)-44 (150 mg, 0.36 mmol) in anhydrous THF (3 mL). After stirring at 20 °C for 15 h,  $H_2O$  (4 mL) was added dropwise, and then MeOH  $(8 \text{ mL})$  and  $\text{K}_2\text{CO}_3$   $(200 \text{ mg})$  were added. After heating to 60 "C for 2 h, the solvent was evaporated and the residue **was** taken up with AcOEt (10 mL). The precipitate was filtered off and the solvent was evaporated. The residue was purified by column chromatography on silica gel (3 g,  $CH_2Cl_2$ / MeOH, 5:1,  $R_f = 0.25$ ), yielding 95 mg (95%) of a colorless oil:<br>
<sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>)  $\delta$  7.43-7.28 (m, C<sub>6</sub>H<sub>5</sub>), 4.85 (2 d, J<br>
= 11.5, Bn), 4.28 (ddd, J = 4.5, 3.5, 2.0, HC(1)), 3.72 (ddd, J =<br>
10.5, 9.0, 5.0, 10.5, 9.0, 5.0, HC(6)), 3.59 **(t**,  $J = 9.0$ , HC(8)), 3.45 **(t**,  $J = 9.0$ , HC(7)), 3.19 **(dd,**  $J = 10.5$ **, 5, HC(5))**, 3.13 **(td,**  $J = 8.0$ **, 2.0, HC(3)**), 2.29-2.10 (m, HC(2), HC(3)), 1.97 (t,  $J = 10.5$ , H'C(5)), 1.95 (dd,  $J = 9.0, 3.5, HC(8a)$ , 1.86-1.75 (m, H'C(2)).

**(+)-Castanospermine** ((+)-1). A mixture of *45* (84 **mg),** THF (1 **mL),** H20 (0.2 **mL),** and 10% Pd/C (100 *mg)* was degassed and pressurized (1 atm) with  $H_2$ . After stirring at 20 °C for 24 h, the precipitate was fitered off on Celite and rinsed with MeOH. The filtrate was concentrated in vacuo, and the residue was dissolved in H<sub>2</sub>O (6 mL). The aqueous solution was washed with  $CH_2Cl_2$ (3 mL, five times). Each organic layer was extracted with water (4 mL). The aqueous phases were combined and basic Amberlite IRA 68 (500 mg, previously washed with  $H_2O$ ) was added, and the suspension was stirred for 20 min. After fiitration, the solvent was evaporated and the residue was dried in vacuo, yielding *55*  mg (97%) of a colorless oil that crystallized slowly. Recrystallization from EtOH (4 **mL)** yielded 41 *mg* of pure (+)-1: colorless crystals; mp 208-209 °C dec (lit. mp 212-215 °C,<sup>3</sup> 207-210 °C<sup>15</sup>); 1.5, HC(1)), 3.61 (ddd,  $J = 10.5$ , 9.0, 5.0, HC(6)), 3.56 (dd,  $J = 10.0$ , 9.0, HC(8)), 3.29 (t,  $J = 9.0$ , HC(7)), 3.15 (dd,  $J = 11.0$ , 5.0, HC(5)), 3.06 (ddd,  $J = 9.5, 9.0, 2.0,$  HC(3)), 2.29 (dddd,  $^2J = 14.0,$  ${}^{3}J = 7.0, 9.0, 2.0, \text{HC}(2)$ , 2.20 **(td,**  ${}^{2}J = {}^{3}J = 9.0, \text{H}^{\prime}\text{C}(3)$ ), 2.04 (dd,  $^{2}J = 11.0$ ,  $^{3}J = 10.5$ , H'C(5)), 2.01 (dd,  $J = 10.0$ , 4.5, HC(8a)), 1.68 (dddd,  $^2J = 14.0$ ,  $^3J = 9.5$ , 9.0, 1.5, H'C(2)). The same characteristics were observed with a sample of natural **(+)-l**  supplied by Sigma Chemical Company, St. Louis, MO 63178 (C3784): *'3c NMR* **(90.55** *MHz* D20) b 79.4 (d, J(C,H) = 145, C(7)), 71.8 (d, 140, C(8a)), 70.5 (d, 145, C(8)), 70.0 (d, 150, C(l)), 69.3 (d, 145, C(6)), 55.7 (t, 140, C(5)), 51.9 (t, 140, C(3)), 33.1 (t, 135,  $C(2)$ );  $[\alpha]^{20}$ <sub>589</sub> = +81°,  $[\alpha]^{20}$ <sub>578</sub> = +83°,  $[\alpha]^{20}$ <sub>546</sub> = +93.5°,  $[\alpha]^{20}$ <sub>436</sub>  $= +153^\circ$ ,  $[\alpha]^{\frac{20}{386}} = +235^\circ$   $(c = 2.2 g/dm^3)$ ,  $\frac{\text{H}_2\text{O}}{[\text{H}t}$ .  $[\alpha]^{\frac{18}{389}} = +79.7^\circ$  (c = 9.3 g/dm<sup>3</sup>),<sup>3</sup> +71° (c = 2.7 g/dm<sup>3</sup>),<sup>14</sup>  $[\alpha]^{\frac{19}{D}} = +7.9^\circ$  $(c = 2 \text{ g}/\text{dm}^3)^{15}$ ]; MS (70 eV) 189 (M<sup>+</sup>, 19), 172 (10), 171 (10), 149 (ll), 145 (67), 128 (14), 100 **(15),** 99 (13), 98 (15), 86 (1001, 85 (16), *<sup>84</sup>*(ll), 71 (16), 70 (22), 69 (ll), 68 (20), 60 (32), 58 (27), 57 (65), 56 (17), 55 (21), 45 (33). Anal. Calcd for C<sub>8</sub>H<sub>15</sub>NO<sub>4</sub> (189.21): C, 50.78; H, 7.99; N, 7.40. Found: C, **50.85;** H, 7.96; N, 7.44. <sup>1</sup>H NMR (360 NMR (360 MHz,  $D_2O$ )  $\delta$  4.38 (ddd,  $J = 7.0, 4.5,$ 

(\*)-Cartanorpermine **((\*)-1).** The same procedure was followed as for  $45$  and  $(+)$ -1, starting with  $(\pm)$ -44: colorless crystals; mp 189-190 "C.

( 1 S ,7R ,8R *,8aR* **)-8- (Ben** zy10xy)- 1 ,7-dihydroxyoctahydroindolizidine  $((+)$ -46). BH<sub>3</sub>·Me<sub>2</sub>S (4 mL) was added dropwise to a stirred solution of **43** (740 mg, 90% pure, 2.43 mmol). After stirring at 20 °C for 4 days, the mixture was cooled to 0 °C, and  $H<sub>2</sub>O$  (15 mL) and then 3 N HCl (15 mL) were added dropwise with stirring. The mixture was heated to 60 "C for 4 h, and the solvent was evaporated. The residue was purified by column chromatography on Dowex-H<sup>+</sup> (20 g). The elution started with 1:5  $\text{MeOH}/\text{H}_2\text{O}$  (60 mL) and continued with  $\text{H}_2\text{O}$  (50 mL), then with MeOH  $(50 \text{ mL})$ , and finally with 1 N NH<sub>3</sub> in 1:1 MeOH/H<sub>2</sub>O. The main fraction gave **480** mg of yellowish oil which was purified by column chromatography on silica gel (30 g,  $CH_2Cl_2/MeOH$ , 1011, yielding 158 *mg* (25%) of a colorless **oil:** 'H *NMFt* (250 *MHz,*  CDCl<sub>3</sub>)  $\delta$  7.42-7.20 (m, 5 H), 4.92 and 4.76 (2 d, <sup>2</sup>J = 12.0, CH<sub>2</sub>Ph),  $4.32 \text{ (m, HC(1))}, 3.56 \text{ (ddd, } J = 11.0, 9.0, 5.0, \text{HC(7))}, 3.50 \text{ (t, } J = 9.0, \text{HC(8))}, 3.15 \text{ (m, } H_{\text{eq}}\text{C(3))}, 3.02 \text{ (ddd, } J = 11.5, \, \sqrt[3]{J} = 4.5,$ 2.5,  $H_{eq}C(5)$ ), 2.30-1.60 (m,  $\overrightarrow{7}$  H,  $H_{ex}C(3)$ ,  $H_{ex}C(5)$ ,  $H_{2}C(6)$ ,  $H_{2}C(2)$ );  $[\alpha]^{25}$ <sub>589</sub> = +3.2°,  $[\alpha]^{25}$ <sub>578</sub> = +3.5°,  $[\alpha]^{25}$ <sub>546</sub> = +3.5°,  $[\alpha]^{25}$ <sub>436</sub> = +4.2°,  $[\alpha]^{25}_{365}$  = +7.1°,  $(c = 6.2 \text{ g}/\text{dm}^3, \text{CH}_2\text{Cl}_2)$ .

(lS,7R,8R,8aR )- **1,7,8-Trihydroxyoctahydroindolizidine (6-Deoxycastanospermine, (+)-2).** A mixture of (+)-46 (158 mg, 0.6 mmol), MeOH (1 **mL), H20** (0.4 mL), HCOOH (0.4 mL), and 10% Pd/C (20 mg) was degassed and then pressurized with H<sub>2</sub> (1 atm.). After shaking at 20 °C for 16 h, the precipitate was filtered off (rinsing with MeOH) and the solvent was evaporated. The residue was dissolved in H<sub>2</sub>O and purified by column chromatography of Dowex 50  $XH^+(2 g)$ . The elution started with matography of Dowex ou An  $(25)$ . The states seems  $\frac{1}{2}$  (97%) of a<br>H<sub>2</sub>O (4 mL), then with 1 N NH<sub>3</sub>, yielding 108 mg (97%) of a  $H_2O$  (4 mL), then with 1 N NH<sub>3</sub>, yielding 106 mg (97%) of a<br>yellowish oil: <sup>1</sup>H NMR (250 MHz, D<sub>2</sub>O)  $\delta$  4.36 (ddd,  $J = 7.0, 5.0$ , 2.0, HC(1)), 3.54-3.44 (m, HC(7), HC(8)), 3.20 (ddd, <sup>2</sup>J = 9.5, <sup>3</sup>J<br>= 9.0, 2.5, H<sub>eq</sub>C(3)), 2.94 (ddd, <sup>2</sup>J = 11.5, <sup>3</sup>J = 4.5, 2.5, H<sub>eq</sub>C(5)),  $2.27 \text{ (ddd, }^2J = 14.0, ^3J = 9.0, 7.0, 2.5, H_aC(2)), 2.12 \text{ (ddd, }^2J = 2.27 \text{ (ddd, }^2J = 14.0, ^3J = 9.0, 7.0, 2.5, H_aC(2)), 2.12 \text{ (ddd, }^2J = 2.27 \text{ (dad, }^2J = 2.0, 7.0, 2.5)$  $3J = 9.0$ ,  $H_{ax}C(3)$ ), 2.08 (dd,  $3J = 9.0$ , 5.0,  $HC(8a)$ ), 1.93 (dddd, *2J* = 13.0, <sup>3</sup>*J* = 4.5, 2.5, 2.5, H<sub>eq</sub>C(6)), 1.89 (dddd, <sup>2</sup>*J* = 14.0, <sup>3</sup>*J* = 9.0, 9.0, 2.0, H<sub>b</sub>C(2)), 1.64–1.56 (m, H<sub>ex</sub>C(6)); [ $\alpha$ ]<sup>25</sup><sub>589</sub> = +36°,  $[\alpha]^{25}$ <sub>578</sub> = +37.4°,  $[\alpha]^{25}$ <sub>546</sub> = +40°,  $[\alpha]^{25}$ <sub>436</sub> = +71°,  $[\alpha]^{25}$ <sub>365</sub> = +72°  $(c = 25 \text{ g/dm}^3, \text{EtOH}).$ 

**(1s** ,6R **,7S,8R,8aR)-1,7-Diacetoxy-8-(benzyloxy)-6 fluorooctahydro-5-indolizidin-5-one** ((+)-47). A mixture of

**43** (102 mg, 0.32 mmol), HF $\cdot$ Et<sub>3</sub>N (3 mL), and BEMP on polystyrene (150 mg) was heated to 95 "C for 2 days and then poured into a saturated aqueous NaHCO<sub>3</sub> solution (30 mL). The mixture was extracted with AcOEt (30 mL, four times); the extracts were combined and dried (MgSO4), and the solvent was evaporated. The yellowish residue (150 mg) was mixed with  $Ac_2O$  (0.5 mL), pyridine (0.5 mL), and **2-(dimethy1amino)pyridine** (1 mg) and stirred at 20 °C for 4 days. The solvent was removed by azeotropic distillation with toluene (twice), and the residue was purified by column chromatography on silica gel (4 g, AcOEt). The main fraction was purified by column chromatography  $(SiO<sub>2</sub>, Lobar$ B, AcOEt). After crystallization from  $Et_2O$ , 66 mg (52%) was obtained: colorless crystals; mp 132-138 "C (for the racemate (±)-47: mp 178-179 °C); <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  7.40-7.22 (m, 5 H), 5.62 (ddd, 3J(H,F) = 14.5, *3J* = 8.5, 9.5, HC(7)), 5.36  $(m, HC(1)), 4.82$  (dd,  $^{2}J(H,F) = 48, {}^{3}J(H,H) = 8.5, HC(6)), 4.65$ = 1.5, HC(8)), 3.75 (dd,  ${}^{3}J(H,H)$  = 9.5, 3.0, HC(8a)), 3.62 (m, H<sub>2</sub>C(3)), 2.19-2.10 (m, H<sub>2</sub>C(2)), 2.13 and 1.97 (2 s, 2 Ac); [ $\alpha$ ]<sup>25</sup><sub>589</sub>  $H_2C(3)$ ), 2.19-2.10 (m,  $H_2C(2)$ ), 2.13 and 1.97 (2 s, 2 Ac); [ $\alpha$ ]<sup>26</sup><sub>589</sub> = +162°, [ $\alpha$ ]<sup>26</sup><sub>436</sub> = +331°,  $[\alpha]^{25}$ <sub>365</sub> = +531°  $(c = 10 \text{ g}/\text{dm}^3, \text{CH}_2\text{Cl}_2)$ . and 4.55 (2 d, <sup>2</sup>J = 11.5, CH<sub>2</sub>Ph), 3.84 (td, <sup>3</sup>J(H,H) = 9.5, <sup>4</sup>J(H,F)

**(1S,6S** ,7S *,8R* **,8aR)-8-(Benzyloxy)-6-fluoro-1,7-dihydroxyoctahydroindolizidine** ((+)-48). A 10 M solution  $BH<sub>3</sub>Me<sub>2</sub>S$  in THF (0.4 mL) was added to a solution of (+)-47 (207 mg, 0.55 mmol) in anhydrous THF (3 mL). After the mixture was stirred at 20 °C for 24 h, 3 N HCl (1 mL) was added dropwise and then MeOH (3 mL) was also added. After the mixture was heated to 70 °C for 4 h, the solvent was evaporated and the residue was dissolved in MeOH  $(3 \text{ mL})$ . NaHCO<sub>3</sub>  $(1 \text{ g})$  was added portionwise, and the mixture was diluted with  $CH_2Cl_2$  (5 mL). The precipitate was filtered off (Celite), and the solvent was evaporated. The residue was purified by column chromatography on silica gel (10 g,  $CH_2Cl_2/MeOH$ , 10:1). The main fraction  $(R_f =$ 0.5) was purified further by column chromatography on silica gel (Lobar B, CH<sub>2</sub>Cl<sub>2</sub>/MeOH 15:1,  $R_f$  = 0.3), yielding 127 mg (83%) of colorless oil: 'H NMR (250 MHz, CDC13) **6** 7.50-7.26 (m, 5 H), 4.88 and 4.84 (2 d, <sup>2</sup>J = 11.5, CH<sub>2</sub>Ph), 4.54 (dddd, <sup>2</sup>J(H,F) = 51,  ${}^{3}J = 14, 8.5, 5.5, \text{HC}(6)$ , 4.30 (m, HC(1)), 3.75 (ddd,  ${}^{3}J(\text{H,F}) =$ 15,  ${}^{3}J = 9$ , 8.5, HC(7)), 3.60 (t,  ${}^{3}J = 9.0$ , HC(8)), 3.33 (ddd,  ${}^{2}J =$  $10.0, {}^{3}J = 5.5, 2.0, \, \text{H}_{eq}C(5)$ ), 3.4 (m,  $\text{H}_{eq}C(3)$ ), 2.33-2.13 (m,  $\text{H}_{ar}C(3)$ )  $HC(2), H_{ax}C(5)$ , 2.00 (dd, <sup>3</sup>J = 9.0, 3.5, HC(8a)), 1.87-1.74 (m,  $H_bC(2)$ );  $[\alpha]^{25}_{.589}$  = +52°,  $[\alpha]^{25}_{.578}$  = +54°,  $[\alpha]^{25}_{.546}$  = +61°,  $[\alpha]^{25}_{.436}$  $= +98^{\circ}, [\alpha]^{25}_{365} = +149^{\circ}$   $(c = 4.2 \text{ g}/\text{dm}^3, \text{CH}_2\text{Cl}_2).$ 

**(1S,6S,7S,8R,8aR)-6-Fluoro-l,7,8-trihydroxyoctahydro**indolizidine **(6-Deoxy-6-fluorocastanospermine,** (+)-3). Same procedure as for  $(+)$ -2, starting with  $(+)$ -48 (127 mg, 0.45 mmol). Yield: 80 mg (93%) of a white solid. Recrystallization from EtOH (0.3 mL) and  $Et_2O$  (3 mL) at -20 °C gave 60 mg of colorless crystals: mp 142-143 °C. For the racemic  $(\pm)$ -3: mp 154-155 OC dec; IR (KBr) **Y** 3360,2980,2960,2935,2910,2800,1475,1435, **1380,1315,1295,1275,1250,1225,1200,1170,1145,1130,1090,**  1075,1055,1000,980,955,855 cm-'; 'H NMR (250 MHz, **DzO)**   $\delta$  4.44 (dddd, 1 H, <sup>2</sup>J(H,F) = 50, <sup>3</sup>J(H,H) = 10.5, 9.0, 5.5, HC(6)), 4.36 (m, HC(1)),  $3.64-3.50$  (m, HC(7), HC(8)),  $3.30$  (ddd,  $^{2}J=10.5$ ,  ${}^{3}J = 5.5, {}^{3}J(H,F) = 2.2, H_{eq}C(D)$ , 3.09–3.04 (m,  $H_{eq}C(3)$ ), 2.37–2.17 1.94-1.83 (m, HC(2)); 18C NMR (62.9 MHz, D\O) **6 x.0** (dd,  ${}^{1}J(C,F) = 174, {}^{1}J(C,H) = 155, C(6), 77.4$  (dd,  ${}^{2}J(C,F) = 17.3$ )  ${}^{1}J(C,H) = 145, C(7)$ , 71.3 (d,  ${}^{1}J(C,H) = 130, C(8a)$ ), 69.6 (d,  $VJ(C,H) = 155$ , C(1)), 68.4 (dd,  $V(C,F) = 11.1$ ,  $V(C,H) = 145$ , C(8)), 52.8 (td,  $V(C,H) = 140$ ,  $V(C,F) = 26.0$ , C(5)), 51.6 (t,  $V(C,H) = 140$ ,  $V(C,H) = 140$ ,  $V(C,H) = 26.0$ , C(5)), 51.6 (t,  $V(C,H) = 140$ 140, C(3)), 33.0 (t,  ${}^{1}$ J(C,H) = 135, C(2)); MS (CI, NH<sub>3</sub>) 193 (10), 192 (M + 1, 100), 191 (15), 190 (lo), 174 (24), 173 (23), 154 (14), 147 (50), 118 (9), 86 (9), 82 (9);  $[\alpha]^{25}$ <sub>589</sub> = +88°,  $[\alpha]^{25}$ <sub>578</sub> = +92°,  $[\alpha]^{25}_{546}$  = +104°,  $[\alpha]^{25}_{436}$  = +170°,  $[\alpha]^{25}_{365}$  = +267°  $(c = 1.6 \text{ g/dm}^3)$ ,  $U = 5.5$ ,  $U(H, F) = 2.2$ ,  $H_{eq}C(D)$ ), 3.09–3.04 (m,  $H_{eq}C(3)$ ), 2.37–2.17<br>(m,  $H_{eq}C(5)$ ,  $H_{eq}C(3)$ ,  $H_{eq}C(2)$ ), 2.03 (dd,  $U = 9.5$ , 4.5, **H**C(8a)), EtOH). Anal. Calcd for  $C_8H_{14}FNO_3$  (191.20): C, 50.25; **H**, 7.38; N, 7.33. Found: C, 50.36; H, 7.37; N, 7.28.

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**Registry No.** (+)-1, 79831-76-8; (±)-1, 123284-48-0; (+)-2, 130948-07-1; **(+)-3,** 131635-62-6; **(+)-3,** 131722-90-2; (-)-5, 94482-75-4; (±)-5, 94482-73-2; (±)-17, 131635-63-7; (±)-18, 131635-64-8; (-)-24, 131722-80-0; (±)-24, 115140-08-4; (+)-25, 131722-81-1; **(f)-25,** 123190-65-8; **(+)-26,** 131697-89-7; **(\*)-26,**  123190-66-9; (-)-27, 131635-65-9; (±)-27, 123190-67-0; (+)-28, 131635-66-0; (f)-28, 123190-63-6; (-)-29, 131635-67-1; **(-)-30,**  131722-83-3; 8-34, 131722-91-3; **(+)-38,** 131722-84-4; **(f)-38,**  123190-72-7; **(f)-39,** 131635-69-3; 40, 131722-85-5; (+)-41, 131722-86-6; 43, 131722-87-7; (+)-44, 131722-88-8; (±)-44, 123190-76-1; 45, 131722-89-9; (+)-46, 131635-70-6; (+)-46 diacetate,  $131635-73-9$ ; (+)-47,  $131635-71-7$ ; (±)-47,  $131722-92-4$ ; (+)-48, 131635-72-8. 131635-68-2; (-)-32, 131724-03-3; **(+)-33,** 131722-82-2; a-34,

Supplementary Material Available: IR, <sup>13</sup>C NMR, and MS spectral data, **as** well **as** elemental analyses of **all** new compounds  $(+)$ -25 to  $(-)$ -30,  $(-)$ -32 to 34,  $(-)$ -38 to  $(+)$ -41, 43 to  $(+)$ -48; X-ray crystallographic results on **(+)-3,** with stereoview of unit cell (16 pages). Ordering information is given on any current masthead page.

## **Ascent of the Aldose Series by Four Carbon Atoms: Total Synthesis of**  D-glycero -D- *tal0* -L- *tal0* **-Undecose Pentaacetonide**

Giovanni Casiraghi\* and Lino Colombo

Dipartimento di Chimica dell'Università, Via Vienna 2, I-07100 Sassari, Italy

Gloria Rassu\* and Pietro Spanu

*Zstituto per l'dpplicazione delle Tecniche Chimiche Avanzate ai Problemi Agrobiologici del C.N.R., Via Vienna 2,Z-07100 Sassari, Ztaly* 

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Enantiomerically pure undecose acetonide **9** was synthesized, through heptose intermediate **5,** starting with D-glyceraldehyde acetonide (1). The key steps were two consecutive four-carbon homologations, each consisting of four reactions: (i) stereoselective elongation of the aldehyde precursor with **2-(trimethylsiloxy)furan,** giving C,,, butenolide templates **2** and **6,** (ii) anti-selective cis-dihydroxylation of the butenolide double bond, giving fully functionalized lactones 3 and **7,** (iii) lactone ring opening and protection, giving open-chain methyl esters 4 and 8, and (iv) DIBAL reduction to aldoses **5** and 9. At the end of the eight-step sequence, undecose **9** was prepared in a 5.1% overall yield, which corresponded to a 69.5% average yield per step.

The elongation of "short" homochiral progenitors by suitable carbon fragments is a prominent method of ascending the carbohydrate series.' Ingenious approaches to the stereocontrolled assembly of natural and synthetic